

# Formation of octatetrayne ( $\text{HC}_8\text{H}$ ) from the reaction of butadiynyl ( $\text{C}_4\text{H}$ ) with butadiyne ( $\text{HC}_4\text{H}$ )

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## Abstract

We investigated the dynamics of reaction  $\text{C}_4\text{H} + \text{C}_4\text{H}_2 \rightarrow \text{C}_8\text{H}_2 + \text{H}$  by interrogating the time-of-flight (TOF) spectra, the laboratory angular distribution, and the photoionization-efficiency (PIE) spectrum of product  $\text{C}_8\text{H}_2$ .

Reactants  $\text{C}_4\text{H}$  and  $\text{HC}_4\text{H}$  were synthesized from 1%  $\text{C}_2\text{H}_2/\text{He}$  in two crossed molecular beams by pulsed high-voltage discharge. We measured the angle-specific time-of-flight spectra and the photoionization-efficiency spectrum of product  $\text{C}_8\text{H}_2$  using a crossed-molecular-beam quadrupole-mass apparatus and synchrotron vacuum-ultraviolet photoionization. The  $\text{C}_8\text{H}_2$  product has translational energy stretching to the energetic limit of producing  $\text{HC}_8\text{H} + \text{H}$  and has an ionization threshold  $9.0 \pm 0.1$  eV in good agreement with that of  $\text{HC}_8\text{H}$ . The quantum-chemical calculations are in accord with the experimental results. This work demonstrates that octatetrayne can be synthesized from the reaction  $\text{C}_4\text{H} + \text{HC}_4\text{H} \rightarrow \text{HC}_8\text{H} + \text{H}$ .

***HC<sub>8</sub>H***

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