

Investigation the reaction behavior of Photocatalyst CO₂ Reduction by verify the graphene oxide surface property.

Yu-Chung Chang,^{1,2,3} Hsiang-Ting Lien,¹ Chia-Hui Wang,⁴ Indrajit Shown,³ Po-Wen Chung,^{4*} Chen-Hao Wang,^{2*}
Kuel-Hsien Chen,^{3*} Li-Chyong Chen^{1*}

1. Center of Condensed Matter Sciences, National Taiwan University, Taiwan
2. Department of Materials and Sciences, National Taiwan University of Science and Technology, Taiwan
3. Institute of Atomic and Molecular Sciences, Academia Sinica, Taiwan
4. Institute of Chemistry, Academia Sinica, Taiwan

Abstract

A key observation by controlling the surface functional group of graphene oxide under N₂ environmental condition with various treated temperature can tune the surface property of graphene oxide. However, the major function of oxygen functional groups still not clear. In this study, as-prepared graphene oxide (as-prepared GO) and modified-graphene oxide (modified-GO) has been presented. Mostly, the oxygen functional groups on the graphene oxide basal plane have been removed above 400 °C, but hydroxyl functional group still bond with carbon on graphene oxide edge side strongly. modified-GO is optimized by the treated temperature which perform higher CO₂RR conversion yield of C1 and C2 products also C3 product (like acetone) during the reaction process. Furthermore, CO₂ and water vapor absorption are one of the important parameters during CO₂ reduction reaction. In this regard, the CO₂ adsorption of GO and modified-GO have been done. Comparing to the GO, modified-GO shows better CO₂ adsorption behavior than GO. It also provides a reference which can proof the dominated functional groups can change the surface chemistry of the carbon based photocatalyst and makes the final CO₂ conversion products different.

Keywords - Photocatalyst CO₂ reduction reaction, graphene oxide, CO₂ adsorption